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Chapter 02 Testbank

Student: _____

- Anything that occupies space and has mass is called
 - an electron.
 - living.
 - matter.
 - energy.
 - space.
- The electrons of an atom are
 - always equal to the number of neutrons in an atom.
 - found in the nucleus.
 - used to determine atomic number.
 - positively charged.
 - moving in pathways called orbitals.
- All of the following pertain to the atom ^{14}C except (X)
 - has 6 protons.
 - has 6 electrons.
 - has 14 neutrons.
 - is an isotope of carbon.
 - mass number is 14.
- The subatomic particles that surround the nucleus are the
 - electrons.
 - protons.
 - neutrons.
 - protons and neutrons.
 - protons and electrons.
- Cations are
 - charged subatomic particles.
 - atoms that have gained electrons.
 - radioactive isotopes.
 - capable of forming ionic bonds with anions.
 - atoms without protons.

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