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Acid Rain Research Paper

Acid Rain

Rain water is naturally acidic because carbon dioxide in the atmosphere combines with water molecules to form carbonic acid. When sulfur dioxide and nitrogen oxides in the atmosphere react with oxygen in the air forming sulfuric acid and nitric acid, acidic precipitation occurs in the form of rain, snow, or dust. Acid precipitation is considered to have a pH of 5.0 or lower. Acid deposition is more commonly used to describe acid precipitation because it includes both wet and dry decomposition.

Discovery of Acid Rain

The first reports of acid rain were in Manchester, England which was an important city during the Industrial Revolution. The relationship between acid rain